

**CHEM 240: Chemical Reactivity II**

The purpose of this syllabus is to describe the course, resources, and policies. It is meant to help all students understand the expectations and requirements for the course, and it should be used as a reference for questions about policies. When updates to the syllabus are made during the term, a new version will be posted electronically, and all students will be notified.

**Course Information**

**Course:** Chemistry 240 – Chemical Reactivity II (3 credits: Lecture & Discussion)

**Prerequisites:** Completion of Chem 180 and Chem 181 with a grade of C- or better.

**Course Description**

This course is the third in a sequence of multiple chemistry courses designed to create foundational knowledge and proficiency in essential chemistry concepts and skills. It includes topics discussing the reactivity of: alkenes, arenes, alkynes, and polyfunctional organic molecules. An understanding of these structures will build upon previous topics: acids and bases, buffers, chemical equilibrium, molecular thermodynamics and kinetics, nucleophilic substitutions, elimination reactions, carbonyl compounds and reactions with applications to biochemical pathways. Historical and current developments in chemistry as well as real-world problems that chemists address are incorporated into the course.

Alongside specific content, these themes will cycle through each of the foundational courses. They include:

- Structure-Activity Relationships
- The culture and practice of science.
- Energy.
- Polymers, proteins, and macromolecules.
- Sustainability.
- Chemical synthesis, purification, characterization, and analysis.

**Learning Outcomes**

The emphasis of this course is on understanding, prediction, investigation, explanation and evaluation over memorization. This means that students must foster their problem solving skills, ability to make claims based on evidence, use and understanding of models and their limitations, and skills of effective communication of scientific results. It is not enough to know *what* happens in chemistry, the student must also be able to explain *why* it happens. When successful, a student will be able to:

- Qualitatively and quantitatively describe and explain how molecular structure and stability changes over time in chemical reactions.
- Draw and interpret multiple representations of structures depicting reactivity.
- Predict and draw reaction products.
- Propose and draw logical reaction mechanisms.
- Quantify relationships between variables controlling chemical systems.
- Differentiate among closely related factors, categorize problem types, and select appropriate tools to solve these problems.
- Apply chemical principles to explain natural phenomena.

**Class Meeting Times and Locations**

Type (Section)	Day & Times (CDT)	Location
Lecture (001)	Tues. & Thurs., 10-11:15 am	Flanner Hall Auditorium
Discussion (002)	Wed., 11:30 am - 12:20 pm	Flanner Hall 105
Discussion (003)	Wed., 12:35-1:25 pm	Flanner Hall 105

**Instructor Information**

Name	Role	Contact Information	Office hours
James Devery, Ph. D.	Instructor	jdevery@luc.edu Flanner Hall 215	Full schedule on <a href="#">Resources for Help (Sakai)</a> ; by drop-in as needed.

**Email Policy**

Direct all emails to [jdevery@luc.edu](mailto:jdevery@luc.edu) with "CHEM 240" as the subject line. In most cases I will be able to respond within 24 hours Monday-Friday when classes are in session. You are encouraged to use Office Hours to get immediate answers to your questions, and to use your classmates as resources for help. You are welcome to email us in the evenings/nighttime, and you can expect a response sometime during the next day.

**Required Resources**

- (1) WileyPlus, see [Sakai](#) for additional information and recommendations
- (2) Electronic resources, e.g., [Loyola Sakai](#) & email, Gradescope
- (3) Molecular model kit
- (4) Scientific calculator

**Copyright/Intellectual Property reminder:** Course materials provided by your instructors at Loyola, including our materials, may not be shared outside any course without the instructor's **written permission**. Content posted without permission will be in violation of Copyright/Intellectual Property laws.

**Pass/Fail Conversion Deadlines and Audit Policy**

A student may request to convert a course into or out of the "Pass/No-Pass" or "Audit" status only within the first two weeks of the semester. For the Fall 2023 semester, students can convert a class to "Pass/No-Pass" or "Audit" through Monday, September 11th. Students must submit a request for Pass/No-Pass or Audit to their Academic Advisor.

**Final Exam**

The University sets the schedule for all final exams, and has posted the schedule for Fall 2023: [link to schedule](#). The final will be held on:

**Tuesday December 12<sup>th</sup>, 1:00-3:00 pm.**

You will have exactly 2 hours to complete the exam. Additional time will not be granted, even if you start late. There will be no make-up final exams given under any circumstance, and the exam will not be given early, either.

Instructors may not reschedule final exams for a class for another day and/or time during the final exam period. There can be no divergence from the posted schedule of dates for final exams. Individual students who have four (4) final examinations scheduled for the same date may request to have one of those exams rescheduled. If a student reports having four final examinations scheduled for the same date, students should be directed to e-mail a petition to Adam Patricoski, Assistant Dean for Student Academic Affairs, CAS Dean's Office ([apatricoski@luc.edu](mailto:apatricoski@luc.edu)).

**Course Repeat Rule**

Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). The Department advises that it is preferable to complete a course with a grade of C or C-, and to demonstrate growth in future coursework, than to withdraw from a course.

After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of

Chemistry & Biochemistry website: <https://www.luc.edu/chemistry/forms/> and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

### **Student Accommodations**

Loyola University Chicago provides reasonable accommodations for students with disabilities. Any student requesting accommodations related to a disability or other condition is required to register with the Student Accessibility Center (SAC). Professors will receive an accommodation notification from SAC, preferably within the first two weeks of class. Students are encouraged to meet with their professor individually in order to discuss their accommodations. All information will remain confidential. Please note that in this class, software may be used to audio record class lectures in order to provide equitable access to students with disabilities. Students approved for this accommodation use recordings for their personal study only and recordings may not be shared with other people or used in any way against the faculty member, other lecturers, or students whose classroom comments are recorded as part of the class activity. Recordings are deleted at the end of the semester. For more information about registering with SAC or questions about accommodations, please contact SAC at 773-508-3700 or [SAC@luc.edu](mailto:SAC@luc.edu).

*If you use the Testing Center, please schedule all of the tests for this class at the beginning of the semester. If a scheduled test date changes, you will still be accommodated if you had scheduled your test in advance.*

### **Academic Integrity**

All students in this course are expected to have read and to abide by the demanding standard of personal honesty, drafted by the College of Arts & Sciences, which can be viewed at:

<https://www.luc.edu/cas/advising/academicintegritystatement/>

A basic mission of a university is to search for and to communicate the truth as it is honestly perceived. A genuine learning community cannot exist unless this demanding standard is a fundamental tenet of the intellectual life of the community. Students of Loyola University Chicago are expected to know, to respect, and to practice this standard of personal honesty.

Academic dishonesty can take several forms, including, but not limited to cheating, plagiarism, copying another student's work, and submitting false documents.

An instance of academic misconduct (including those detailed on the website provided above or in this syllabus) will be reported to the Department Chair and the academic Dean's office. Evidence of cheating in this course will result in, at a minimum, a score of zero (which cannot be dropped from grade calculations) and penalty up to failure of the course.

### **Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC):**

Students missing classes while representing Loyola University Chicago in an official capacity (e.g., intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes.

Students should discuss with faculty the potential consequences of missing lectures and the ways in which they can be remedied. Students must provide their instructors with proper documentation i.e., "Athletic Competition & Travel Letter" describing the reason for and date of the absence.

This documentation must be signed by an appropriate faculty or staff member and it must be provided to the professor in the first week of a semester. It is the responsibility of the student to make up any assignments. If the student misses an examination, the instructor is required to allow the student to take the examination at another time.

(<https://www.luc.edu/athleteadvising/attendance.shtml>)

Students who will miss class for an academic competition or conference must provide proper documentation to their instructor as early in the semester as possible.

**Accommodations for Religious Reasons**

If you have observances of religious holidays that will cause you to miss class or otherwise effect your performance in the class you must alert the instructor **within 10 calendar days of the first class meeting of the semester** to request special accommodations, which will be handled on a case by case basis.

**Universal Absence Accommodation Policy**

The purpose of a universal absence accommodation policy is to account for emergency circumstances (e.g., serious illness, caring for a family member, car accident) that require you to be absent from class, while maintaining fairness in grading for students who attend and complete all in-class graded assignments. We believe that class attendance and participation are essential for your success in this class, and that your health is important to us and our shared community. Please use good judgement and stay home if necessary/prudent for your circumstances.

This is the universal accommodation policy for in-class graded assignments:

- Group Quizzes: the specification for an A is 9 to allow for missed work.
- COs: you are eligible to submit for Proficiency after the first attempt at an CO whether you complete the problems or not; reattempts at Mastery are available during the term.

You may provide documentation for an absence, but it is not required. These accommodations are automatically available to all students.

**Class Recording & Content Information**

In general lecture, meetings may be recorded. The following is a mandatory statement for all courses in the College of Arts & Sciences (CAS). We will discuss class norms and standards during the first week and continue the discussion as needed throughout the semester.

**Privacy Statement**

Assuring privacy among faculty and students engaged in online and face-to-face instructional activities helps promote open and robust conversations and mitigates concerns that comments made within the context of the class will be shared beyond the classroom. As such, recordings of instructional activities occurring in online or face-to-face classes may be used solely for internal class purposes by the faculty member and students registered for the course, and only during the period in which the course is offered. Students will be informed of such recordings by a statement in the syllabus for the course in which they will be recorded. Instructors who wish to make subsequent use of recordings that include student activity may do so only with informed written consent of the students involved or if all student activity is removed from the recording. Recordings including student activity that have been initiated by the instructor may be retained by the instructor only for individual use.

**Additional Content, Copyright & Intellectual Property Statement**

By default, students may not share any course content outside the class without the informed written consent of the owner of that content. This includes any additional recordings posted by students, materials provided by the instructor, and publisher-provided materials. For example, lectures, quiz/exam questions, book figures/slides, and videos may not be shared online outside the class. In some cases, copyright/IP violations may overlap with breaches of academic integrity. Remember that obtaining consent to share materials is an active process.

**Other Items**

- A link to the official Loyola calendar can be found here:  
<https://www.luc.edu/academics/schedules/>
- The Withdraw deadline for the semester is on Friday, November 3.
- Loyola is using SmartEvals to provide instructor & course feedback. OIE will send emails near the end of the term.

### Standards

The standards for each letter grade are listed here according to all required course components, listed in columns. You must meet or exceed all of the standards listed to earn the corresponding letter grade: standards are not averaged across components. These lists are intended for complete transparency: you do not need to do any extra work to figure out what is required for any grade, and we will revisit the standards and expectations after the early rounds of testing to help you gauge your progress in the course. Grades are only based on the criteria listed in the syllabus: no substitutions, and no additions. Descriptions of the components are found on the preceding pages.

	CO Mastery				CO Total (Mastery + Proficiency)	Reaction Writing	Quizzes	WileyPlus
	Synthesis	Mechanism	Spectroscopy	Total Mastered				
A	≥7	≥7	≥7	≥22	≥23	≥3	≥9	≥90
A-	≥6	≥6	≥6	≥20	≥23	≥3	≥9	≥90
B+	≥5	≥5	≥5	≥18	≥21	≥2	≥9	≥80
B	≥4	≥4	≥4	≥16	≥21	≥2	≥9	≥80
B-	≥3	≥3	≥3	≥14	≥19	≥2	≥8	≥80
C+	≥2	≥2	≥2	≥12	≥19	≥1	≥8	≥70
C	≥1	≥1	≥1	≥10	≥17	≥1	≥8	≥70
C-	≥1	≥1	≥1	≥6	≥16	≥1	≥7	≥70
D	0	1	2	≥3	≥9	≥0	≥5	≥50

Note: a student who fails to meet the standards for a grade of D will receive a grade of F for the course.

### Course Grading System Design

There are three basic principles that we have used to design the grading system for this course. These are for you to:

1. Understand what the standards and requirements are for each letter grade so that you can choose what level of academic achievement to pursue in this course. We encourage each of you to strive for high achievement because we believe in the potential of all students to learn and improve their abilities in chemistry.
2. Expect a challenging but flexible learning environment. The standards for demonstrating your Mastery of the course material are high in each area, but the methods for meeting the standards are designed to give you multiple chances to revise and improve the quality of your work throughout the semester.
3. Learn from mistakes. Deep, connected learning involves hard work and reflection on your progress. Chemistry is a cumulative subject where the new topics build on prior knowledge and this system is designed for cycles of learning.

### Course Assessment

All of the following are required components of your course grade:

#### Comprehensive Objectives: Mastery Testing & Proficiency Revisions

The purpose of testing is to align your course grade with your level of learning, based on your mastery of comprehensive topics. The purpose of COs is to allow you to demonstrate your higher-level skills of applying and analyzing, requiring you to go beyond memorization of facts and processes and transfer your understanding of essential course concepts to new scenarios. The COs are all related to the Course Content & Learning Outcomes on the first page of this syllabus. A list of COs will be updated for each unit as we progress through the material. COs will be scored as Mastered or Not

Mastered. A score of Mastered is earned for correctness and completeness of the problem(s). Note that the standards for earning Mastery will be high: by definition there is no partial credit, but you will learn the standards from the examples for class activities. Each round of testing on COs will be followed by an opportunity to resubmit work to earn a score of Proficient for an CO that was Not Mastered in the first testing opportunity. Resubmissions for Proficiency will also earn reattempts of COs. Reattempts will take place with the next round of testing. Note that your grade will not count both Mastery and Proficiency for the same item; a CO that is scored Proficient and then is subsequently Mastered on a re-attempt will count only as being Mastered. Rounds of testing are scheduled for **September 14, October 19, November 16, and December 7**, with an additional round scheduled during the final exam period. Specific CO dates and timing will be announced at least one week in advance. All procedures, allowed resources, and requirements will be posted before each round of testing.

### Reaction Writing

Each Unit, you will be asked to write a guide for a reaction from the Unit for your group. The guide will explain how to propose a mechanism for this reaction. The guide will also explain how to use the reaction in multi-step synthesis. The guide will be no more than two pages. Your group will peer review the guide, offering comments for revision. You will be given the opportunity to submit revisions. Upon the acceptance of revisions by your group, they will be submitted for final check by Devery.

### Group Quizzes

On average, 1 quiz per week, usually completed in assigned groups. Most assignments will be completed in class and submitted to Gradescope. The purpose of participation is to improve your learning by: 1) cooperation, communication and support among your classmates as you practice the skills required for success in the course; and 2) providing feedback on your progress to encourage reflection and improvement. Quizzes will include test questions from previous semesters. You will get as much benefit from these quizzes as you choose to put forth in your effort and you are expected to correct your work after receiving feedback. Each quiz will contribute equally toward this category in your course grade. Refer to the Universal Absence Accommodation Policy for missed quizzes.

### WileyPlus: Required Homework

Registration information is on Sakai; use of this system includes eText access. The purpose of these assignments is to help you keep up with the course material by preparing ahead for each class. You will get as much benefit from these assignments as you choose to put forth in your effort to solve the problems on your own: a list of reading (textbook sections) will be continually updated to correlate with the WileyPlus pre-lecture assignments. Typically, you will have 2-4 required assignments per week, always due at 11:59pm, posted at least 48 hours in advance. Assignments will be submitted completely online with the individual grading policy listed with each assignment. Contact [Wiley Support](#) for help with technical aspects of using WileyPlus. Additional practice assignments will be posted that will not count toward the point total for your course grade.

### Posting of Grades

Final course grades at the end of the semester are posted only LOCUS. Grades are never sent via email. WileyPlus scores are automatically recorded in the WileyPlus Gradebook for that system. Scores for all other required components will be made available on Sakai. Each student will see an estimated midterm grade in LOCUS before the withdraw deadline.

### Changes to Syllabus

There may be changes to the syllabus during the semester. ***You are responsible for all syllabus changes made in class whether or not you attend.***